

Vapor-Liquid-Solid Growth of Bi₂O₂Se Nanoribbons for High-Performance Transistors

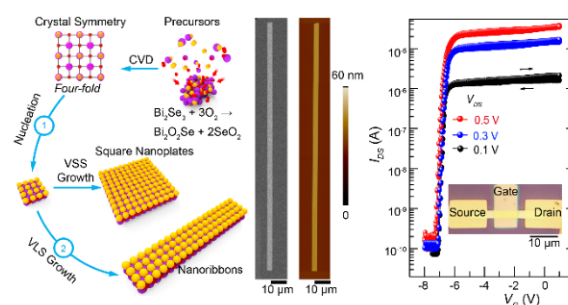
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Abstract: Nanostructured bismuth oxyselenide (Bi₂O₂Se) semiconductor, a two-dimensional (2D) materials with high-mobility, air-stability, and tunable bandgap, has recently emerged as a candidate of channel material for future digital (electronic and optoelectronic) applications. In terms of material morphology, some basic issues will be addressed when a two-dimensional layered crystal is shaped into a one-dimensional (1D) geometry due to size effect; these include the space-confined transport in a plane, which leads to dramatic changes in electronic, optical, and thermal properties. These novel 1D nanostructures with unique properties are an optimal choice for fabricating next-generation integrated circuits and functional devices within the nanometer scale such as gate-all-around field-effect transistors, single-electron transistors, chemical sensors, and THz detectors. As one of the high-mobility 2D semiconductor, 1D high-quality Bi₂O₂Se nanoribbons could be promising for applications in high-performance transistors; however, their synthesis has not been completely developed yet. In our study, we report on the facile growth of Bi₂O₂Se nanoribbons on mica substrates *via* a bismuth-catalyzed vapor-liquid-solid (VLS) mechanism. The preparation of Bi₂O₂Se nanoribbons is based on a previous work that emphasized on the oxidation of Bi₂Se₃ in a chemical vapor deposition (CVD) system and the use of bismuth (Bi) particles as the precursor of Bi catalysis. The morphology, composition, and structure of the as-grown Bi₂O₂Se nanoribbons were characterized by scanning electron microscopy (SEM), atomic force microscopy (AFM), Raman spectroscopy, transmission electron microscopy (TEM), as well as other methods. For a Bi mediated VLS growth process, the growth of Bi₂O₂Se nanoribbons can be self-assembled; further, in this process, as-grown epitaxial Bi₂O₂Se nanoribbons are free-standing with out-of-plane morphology on the mica substrate. Additionally, combining the spherical aberration corrected transmission electron microscope (ACTEM) and selected electron diffraction (SAED) methods, we discovered that the as-synthesized Bi₂O₂Se nanoribbons were single crystalline with high quality. We further investigated the controllable growth for domain size by optimizing the growth temperature of the Bi₂O₂Se nanoribbons. As-synthesized single-crystal Bi₂O₂Se nanoribbons have widths in the range of 100 nm to 20 μm and lengths in the sub-millimeter range. By employing a polymer poly(methyl methacrylate) (PMMA) assisted clean transfer method with the assistance of deionized water, the Bi₂O₂Se nanoribbons can be easily transferred onto a SiO₂/Si substrate. Fabricated into the top-gated field-effect device, the Bi₂O₂Se nanoribbon sample (transferred to the SiO₂/Si substrate) exhibited high electronic performances; these included a high electron mobility of ~220 cm²·V⁻¹·s⁻¹ at room temperature, good switching behavior with on/off ratio of >10⁶, and



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high on current density of $\sim 42 \mu\text{A}\cdot\mu\text{m}^{-1}$ at a channel length of 10 μm . Therefore, $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbons are expected to be a promising materials for building high-performance transistors in the future.

Key Words: Bismuth oxyselenide; Vapor-liquid-solid growth; Nanoribbons; CVD; High mobility

$\text{Bi}_2\text{O}_2\text{Se}$ 纳米带的气-液-固生长与高性能晶体管的构筑

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摘要: 作为一种具有高迁移率、高空气稳定性和带隙可调的二维材料, 纳米硒氧化铋($\text{Bi}_2\text{O}_2\text{Se}$)半导体有望成为未来电子学集成器件和光电子集成器件沟道材料的候选半导体。高质量的 $\text{Bi}_2\text{O}_2\text{Se}$ 纳米带有望用于高性能晶体管的构筑; 然而, 其一维结构的合成方法尚未开发。在我们的研究中, 我们在云母衬底上通过Bi催化汽-液-固生长机制合成了一维 $\text{Bi}_2\text{O}_2\text{Se}$ 纳米带。合成的 $\text{Bi}_2\text{O}_2\text{Se}$ 单晶纳米带的宽度为100 nm到20 μm , 长度可达亚毫米。再者, $\text{Bi}_2\text{O}_2\text{Se}$ 纳米带可以很容易地利用洁净转移方法被转移到 SiO_2/Si 衬底上, 并进一步制备成高性能场效应器件。 $\text{Bi}_2\text{O}_2\text{Se}$ 纳米带场效应器件表现出优异的电学性质: 室温电子迁移率高达 $\sim 220 \text{cm}^2\cdot\text{V}^{-1}\cdot\text{s}^{-1}$, 开关比高达 $> 10^6$, 10 μm 沟道长度下电流密度高达 $\sim 42 \mu\text{A}\cdot\mu\text{m}^{-1}$ 。由此说明, $\text{Bi}_2\text{O}_2\text{Se}$ 纳米带有望成为候选材料用于未来高性能晶体管的构筑。

关键词: $\text{Bi}_2\text{O}_2\text{Se}$; 气-液-固生长; 纳米带; 化学气相沉积; 高迁移率

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1 Introduction

Fabricating advanced technological-node transistors and manipulating electronic properties of two-dimensional (2D) materials hold significance for their future applications in nanoelectronics, for which converting a two-dimensional to one-dimensional (1D) structure (nanowires, nanotubes, and nanoribbons) is an effective strategy. For future technology nodes, gate-all-around field effect transistors (FETs) fabricated on a 1D semiconductor nanostructure are promising candidates to replace the fin-field effect transistors (FinFETs) Fin Field Effect Transistors and planar short-channel FETs owing to a better electrostatic control of the channel transport and facilitating further reductions in transistor size with low leakage currents *via* fully surrounding gate^{1,2}. As the technology node requires to control channel diameter in the sub-3 nm range, except for 2D semiconductor with 1D geometry, most channel materials (Si, Ge and III-V) face process, mobility or quantum capacitance challenges of such ultra-thin body thickness^{3,4}. Besides, some basic issues will be addressed as 2D layered crystal shape into a 1D geometry, such as space-confined transport in a plane and leads to dramatic changes in electronic, optical, and thermal properties^{1,2,5}. Graphene nanoribbons, as a typical example, varies from pristine 2D graphene nanosheets and exhibits non-zero bandgap useful for room temperature transistor operations with excellent switching speed and high carrier mobility^{4,6,7}. Besides, many other ribbon-like 2D

materials, such as MoS_2 ⁸, phosphorene^{9,10}, WSe_2 ², and Bi_2Se_3 ¹¹ nanoribbons, also lead to exceptional control over electronic structure, by which the novel quantum phenomena and unique electric properties can be observed. In this regard, fabrication of 2D materials into 1D nanoribbons (varying from the 2D) is capable of arousing scientific and technological interests.

In terms of 2D materials, a currently arising member, bismuth oxyselenide ($\text{Bi}_2\text{O}_2\text{Se}$), joins the family of layered 2D materials and emerges as a promising candidate for future electronic and optoelectronic applications¹²⁻¹⁵. $\text{Bi}_2\text{O}_2\text{Se}$ has been demonstrated to be of among remarkable characteristics, such as air stability against oxidation and moisture¹², tunable bandgap with the thickness down to a few atomic layers^{12,16}, high electron mobility ($> 20000 \text{cm}^2\cdot\text{V}^{-1}\cdot\text{s}^{-1}$ at 1.9 K)^{12,16-18}, being accessible to single crystal films on a wafer-scale¹⁹. In addition, $\text{Bi}_2\text{O}_2\text{Se}$ exhibits high sensitivity, ultrafast broadband photoresponse (0.3–1.7 μm of wavelength)¹³, and ultrabroadband saturable absorption for the mid-infrared (5.0 μm)²⁰. Interestingly, a strain may even induce the giant polarizations, resulting in piezoelectricity and ferroelectricity of $\text{Bi}_2\text{O}_2\text{Se}$ ²¹. These attractive properties make 2D $\text{Bi}_2\text{O}_2\text{Se}$ became a candidate for creating future infrared photodetector and high-performance electronic devices. Therefore, similar with other 2D materials, it is highly motivated that synthesizing 1D $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbons for realizing potential applications of $\text{Bi}_2\text{O}_2\text{Se}$ among high-performance transistors, but the reliable preparation of $\text{Bi}_2\text{O}_2\text{Se}$

nanoribbons with chemical method remains challenging.

Here, *via* facial bismuth (Bi)-catalyzed vapor-liquid-solid (VLS) growth mechanism, we present a chemical vapor deposition (CVD) method to synthesize high-quality Bi₂O₂Se nanoribbons. In this mean, Bi is used as catalyst to induce the orientated growth of the Bi₂O₂Se, especially at out-of-plane, resulting in formation of one-dimensional nanoribbons. The bottom-up synthesized Bi₂O₂Se nanoribbons have typical width down to ~100 nm, thickness down to 5 nm and length up to 200 μm. A polymer assisted clean transfer method was developed to transfer the Bi₂O₂Se nanoribbons onto the SiO₂/Si substrate. Fabricated into the FET, the Bi₂O₂Se nanoribbon sample (transferred to the SiO₂/Si substrate) exhibited high electronic performances: high electron mobility of ~220 cm²·V⁻¹·s⁻¹ and large current on/off ratios of > 10⁶.

2 Experimental and computational section

2.1 Synthesis of Bi₂O₂Se nanoribbons

The preparation of Bi₂O₂Se crystal is based on previous work on the oxidation of Bi₂Se₃ in CVD system¹⁶. Here, we further extend this methodology to prepare Bi₂O₂Se nanoribbons *via* Bi-catalyzed VLS growth. The Bi₂O₂Se nanoribbons were synthesized by the home-made double-zone CVD system (Thermo Inc.), which equipped with a 12-inch-long and 30-mm-diameter quartz tube. Typically, Bi particles (Alfa Aesar, 99.999%) were placed in the upstream zone, and Bi₂Se₃ bulks (Alfa Aesar, 99.999%) were placed in the second zone. The freshly cleaved mica substrates were placed on top of Bi₂Se₃ bulks with a gap of ~3–5 mm. The mixed carrier gas was high-purity Ar/O₂ gas with typical flow rate of 100 sccm/30 ppm (sccm: standard-state cubic centimetre per minute, 1 ppm = 1 × 10⁻⁶ (volume fraction)), and the pressure of the system was kept at 400 Torr (1 Torr = 133 Pa). The growth range was about 590–620 °C.

2.2 Characterization of Bi₂O₂Se nanoribbons

The morphology of the Bi₂O₂Se nanoribbons was characterized by OM (Olympus DX51 microscope), AFM (Bruker icon), and SEM (Hitachi S4800 field emission). The structure and crystallinity of as-grown Bi₂O₂Se nanoribbons was performed using transmission electron microscopy (TEM, FEI Tecnai F30 and FET Titan Themis G2 operating at 300 kV) with energy-dispersive X-ray (EDX) mapping capabilities. Samples for TEM characterization were transferred onto the carbon film supported gold grids by a polymethyl methacrylate (PMMA)-mediated transfer method with the assistance of deionized water²². Raman spectra and mapping was performed at a wavelength of 633 nm on Scanning near-field Raman spectrometer (Witec RSA300+ optical microscope).

2.3 Fabrication of Bi₂O₂Se-nanoribbon-based FET device

To build top-gate FETs, the Bi₂O₂Se nanoribbons were transferred onto 300 nm SiO₂/Si substrates with the location markers using PMMA-mediated transfer method under the

assistance of deionized water. The electron beam lithography (EBL, FEI Inc.) was adopted to pattern the electrodes (source, drain, and top-gate electrode) in two steps. Firstly, the source and drain electrodes were patterned with standard EBL process, followed with the metal deposition (Pd/Au, 5 nm/40 nm) by thermal evaporation. Secondly, the top-gated electrodes required a second EBL exposure to exploit the ‘window’ for the deposition of the high-κ top-gate dielectric HfO₂ (20 nm) by atomic layer deposition (ALD), then 5 nm/40 nm Pd/Au films were deposited as the top-gate electrode. The as-fabricated Bi₂O₂Se-nanoribbon-based top-gate FETs were measured under ambient conditions on a semiconductor analyzer (Keithley, SCS-4200) combined with micromanipulator 6200 probe station at room temperature.

3 Results and discussion

Upon the Vapor-solid-solid (VSS) growth mode, various synthesis process for Bi₂O₂Se generally tends to form 2D structure (square nanoplates) with a characteristic crystal shape^{12,16,22–25} (Fig. 1a). In VSS process, determined by the inherent free energy of crystal edges and surface diffusion kinetics, the gas/vapor phase precursors are converted to solid-state products to form the in-plane 2D nuclei *via* surface adsorption on the substrate, resulting in the formation of the native 2D layer structure^{8,26}. To prepare 1D morphology, VLS growth is effective mechanism to guide the directed growth of nanostructure, in which 1D nanostructures are synthesized by precipitation from supersaturated catalytic liquid droplets. To this end, we conducted Bi-catalyzed CVD approach to realize VLS growth mode to synthesize the 1D Bi₂O₂Se nanoribbons using Bi₂Se₃ and O₂ as precursors (see Experimental for details). Particularly, there are two advantages for using Bi as a catalyst: (1) Bi acts as a catalyst to realize the VLS growth mechanism for preparation of Bi₂O₂Se nanoribbons^{27–30}; and (2) Bi still acts as a reaction precursor to achieve rapid growth of Bi₂O₂Se nanoribbons according to the thermodynamic phase diagram of Bi₂O₂Se¹⁶.

As shown in Fig. 1b, the as-grown crystal shows a ribbon-like

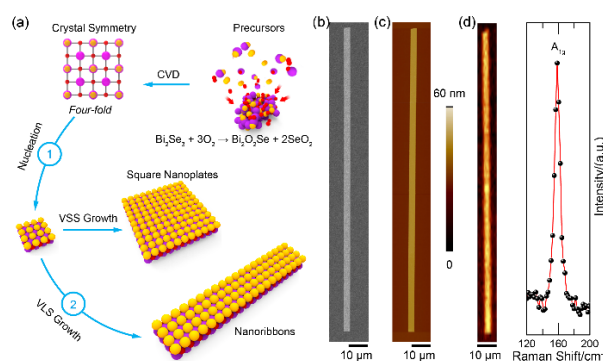


Fig. 1 One-step bottom-up synthesis of Bi₂O₂Se nanoribbons.

(a) Schematic showing growth of Bi₂O₂Se crystals (nanoplates and nanoribbons). (b–d) Scanning electron microscopy (SEM, b), atomic force microscopy (AFM, c), Raman spectra and corresponding mapping (d) images of as-synthesized Bi₂O₂Se nanoribbons.

morphology with the length of sub-millimetre. The atomic force microscopy (AFM) characterization reveals that the surface of the as-grown sample is clean and homogeneous, and its thickness is around 10 nm with a width of 2 μm . Furthermore, Raman spectroscopy indicates that the characteristic A_{1g} peak of $\text{Bi}_2\text{O}_2\text{Se}$ located at $\sim 159\text{ cm}^{-1}$, which is good consistent with the prior reports¹⁵, confirming that as-synthesized nanostructures are $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbons. In addition, the Raman mapping of A_{1g} peak is very uniform across the whole nanoribbon, demonstrating the high uniformity of as-synthesized $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbons.

To study the phase purity and crystalline nature of as-grown $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbons, we performed characterization of transition electron microscopy (TEM). The as-synthesized $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbons were transferred onto the holey carbon-supported Au grid for TEM characterization *via* a polymethyl methacrylate (PMMA)-mediated method. Low-magnitude TEM image shows that the $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbon has a uniform width along the entire length with typical widths of 500 nm (Fig. 2a). As shown in Fig. 2b, the selected-area electron diffraction (SAED) pattern reveals single set of diffraction peaks with the four-fold symmetry, verifying that it is a single crystal in as-grown $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbon. High-resolution TEM image of the $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbon with a schematic of the atomic positions shows well defined lattice spacing of 0.38 nm, which correspond to the theoretical value of lattice for (100) planes in $\text{Bi}_2\text{O}_2\text{Se}$ ¹⁵. Besides, no obvious defects, such as vacancies, interstitials, and dislocations are observed, suggesting that as-grown $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbons have high crystallinity and purity. As characterized

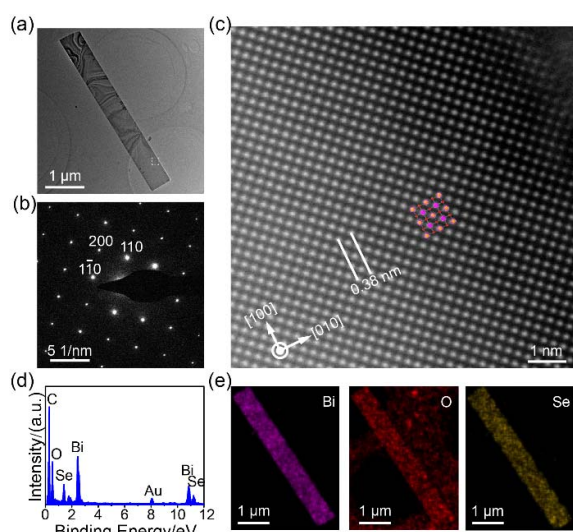


Fig. 2 Structural and crystalline quality characterization of $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbons.

(a) Low-magnitude TEM image of a $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbon that transferred onto the holey carbon-supported Au TEM grid. (b) SAED pattern of as-grown $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbons. (c) Atomically HRTEM images of as-grown $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbons with a schematic of the atomic positions. (d) The EDX spectra of as-synthesized nanoribbons, exhibiting a stoichiometry of $\text{Bi}_2\text{O}_2\text{Se}$. The Au and C signal comes from the TEM grid. (e) The elemental maps of Bi, Se, and O of as-grown nanoribbons.

by energy-dispersive X-ray spectrometry (EDS), the obvious signal peaks for Bi, O, and Se were observed (Fig. 2d) without other signal of impurities, the element ratio of Bi, O, Se was about 2 : 2 : 1, which is in accordance with the stoichiometry of $\text{Bi}_2\text{O}_2\text{Se}$. In addition, corresponding elemental mapping analyses (Fig. 2e) further indicates the uniformity without any impurities, suggesting that the $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbons has good stoichiometry and high purity.

Similarly, some reports have been demonstrated on the catalysis approach for the VLS growth of nanoribbons and nanowire using Bi as catalyst^{27–30}. As shown in the scanning electron microscopy (SEM) images (Fig. 3a), as-synthesized $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbons is out-of-plane on the mica substrate, indicating a Bi mediated VLS growth process. Notably, together with TEM analysis, almost no Bi droplets are found at the tip of $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbons, this is owing to the complete evaporation of Bi of the surface during the cooling process^{27,28}. Interestingly, although the external perturbation is slight, long and thin out-of-plane $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbons can be easily broke from their root. As indicated in Fig. 3b, c, the as-synthesized thin $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbons quickly broken from their root under the low-voltage SEM irradiation (the voltage is 1 kV). Accordingly, it can be inferred that the nanoribbons on the substrate surface are formed by the fracture of thin out-of-plane $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbons due to the perturbation of cooling process (Fig. 1b and 3d).

The ability to tailor the diameter of $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbons for specific applications is essential for synthetic chemistry. In synthesis of $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbons, varying growth temperature allows considerable control over the diameter of the products. As show in Fig. 3d, the length of as-grown $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbons reached a large value of $\sim 280\ \mu\text{m}$ as temperature varied from

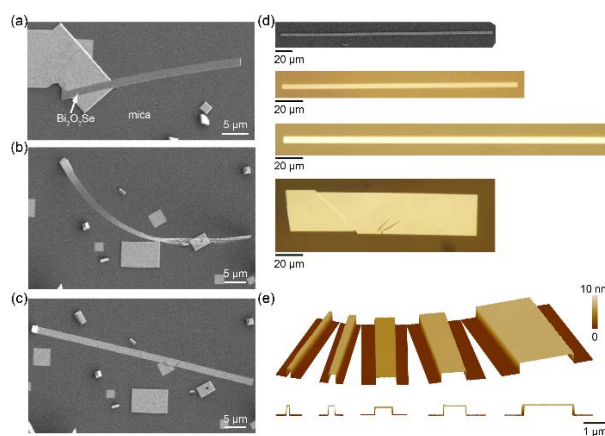


Fig. 3 VLS growth of $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbons on mica substrate.

(a) SEM view images of out-plane $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbons. (b, c) *In situ* SEM images for thin $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbons before and after deformation under the electron beam irradiation. (d) Optical images of as-grown $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbons controlled by the growth temperature, which corresponding to 590 °C (top), 600, 610, 620 °C (bottom), respectively. (e) AFM images of 5 μm long sections of $\text{Bi}_2\text{O}_2\text{Se}$ nanoribbons with widths corresponding to 115 nm, 300 nm, 800 nm, 1.5 μm , 2.4 μm , respectively. Top, 3D view; bottom, the same dataset with the cross view direction through the ribbon length.

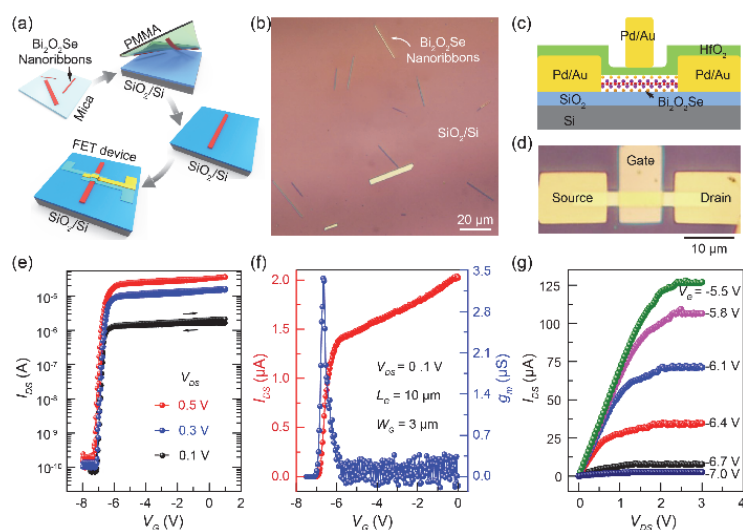


Fig. 4 Fabrication and characterization of FET transistor based on Bi₂O₂Se nanoribbons.

- (a) Schematic of fabrication of FET devices *via* the PMMA-assisted transfer method. (b) Optical microscopy image of as-transferred Bi₂O₂Se nanoribbons on SiO₂/Si substrate. (c) Schematic of the top-gate FET devices. (d) Optical image showing an as-fabricated FET with width (W) of 3 μm and gate length (L_G) of 10 μm . (e) Transfer characteristics for drive voltage (V_{DS}) varying from 0.1 to 0.5 V. (f) Transconductance (g_m) and drain current (I_{DS}) as a function of gate voltage (V_G) curves. (g) Output characteristics of FET curves, showing a large on current density (I_{DS}/W) of $\sim 42 \mu\text{A}\cdot\mu\text{m}^{-1}$.

590 °C to 610 °C, and it shrank gradually as further increasing the growth temperature. The approximate reason for the above phenomenon can be summarized as follows. There are two other factors that affect the Bi-catalyzed VLS growth: the low vapor pressure of Bi and the rapid reaction of Bi in the presence of oxygen and selenide (the reaction is $4\text{Bi}(\text{l}) + \text{Se}_2(\text{g}) + 2\text{O}_2(\text{g}) \rightarrow 2\text{Bi}_2\text{O}_2\text{Se}(\text{s/l})$). When the growth temperature was relatively low, the absorption rate of the Bi-catalyst on the tip of Bi₂O₂Se nanoribbons is relatively high, the catalyzed effect is dominated in VLS process, leading to growth of long and thin nanoribbons. As elevate to high temperature, the evaporation and chemical reaction of Bi is superior to catalyzed effect, the absorption rate of Bi-catalyst diminishes and the catalyzed effect gradually becomes the dominating elemental step for the growth, resulting in wider and shorter of Bi₂O₂Se nanoribbons. Notably, naturally folded without breaking in Bi₂O₂Se nanoribbons can occasionally be formed (Fig. 3d), suggesting that ultrathin nanoribbons are rather flexible. Therefore, upon the aforementioned analysis, as shown in Fig. 3e, we can reliably obtain the thin nanoribbons with different widths and thickness of 5 nm by controlling the growth time under the low temperature (590 °C).

Generally, the direct transfer of Bi₂O₂Se nanoribbons on SiO₂/Si substrates would enable their wide-ranging applications in photonics or electronics because almost all integrated circuits are rooted in the silicon substrate. Hence, as-grown Bi₂O₂Se nanoribbons were firstly transfer onto 300 nm SiO₂/Si substrates to fabricate the FET devices (Fig. 4a). As illustrated in Fig. 4b, as-synthesized Bi₂O₂Se nanoribbons can be easily transfer on SiO₂/Si substrates with various width from Mica substrates, which allow us to evaluate their electronic properties conveniently.

To identify the electronic properties of the nanoribbons, a top-gated device configuration was adopted to evaluate the switching behavior of Bi₂O₂Se nanoribbons with a thickness of 15 nm (Fig. S1, Supporting Information). The structure diagram of Bi₂O₂Se-nanoribbons-based FET devices, and corresponding optical image showing an as-fabricated FET device with a gate length (L_G) of 10 μm and width (W) of 3 μm is shown in Fig. 4c, d, respectively. As shown in Fig. 4e, the measured room-temperature transfer characteristics verify a n-type transistor behavior with a high current on/off ratio (I_{on}/I_{off}) of $\sim 10^6$ at low bias (the level of I_{on}/I_{off} is $> 10^4$ for practical logic transistors) and small subthreshold slope (SS) of $\sim 75 \text{ mV}\cdot\text{dec}^{-1}$. Based on the equation $\mu_{\text{FET}} = \frac{L}{W C_G V_{DS}} \frac{dI_{DS}}{dV_G}$, where L is the channel

length (here, L equal to L_G), W is the channel width, C_G is the capacitance between the channel and the top-gate per unit area ($C_G = \epsilon_0 \epsilon_r / d$, ϵ_0 is vacuum permittivity, ϵ_r is the relative permittivity, and d is the thickness of HfO₂ layer), the field-effect mobility of Bi₂O₂Se-nanoribbons-based FET devices could be calculated by fitting the linear region in transfer curve (Fig. 4f). The field-effect mobility was extracted as $\sim 220 \text{ cm}^2\cdot\text{V}^{-1}\cdot\text{s}^{-1}$, which is the same order as the reported value in a Hall mobility^{11,18}. Besides, the output curves plotted in Fig. 4g shows that the high current density (I_{DS}/W) of $\sim 42 \mu\text{A}\cdot\mu\text{m}^{-1}$ can be achieved. These electronic properties, together with its easy transfer, make Bi₂O₂Se nanoribbons a promising semiconductor candidate for future low-power logic transistors.

4 Conclusions

In conclusion, we have demonstrated the VLS growth of Bi₂O₂Se nanoribbons on Mica substrate *via* Bi-catalyzed CVD. The VLS growth mode is probably conducted by the formation of a Bi liquid solution on the tip of Bi₂O₂Se nanoribbons. The

width and length of Bi₂O₂Se nanoribbons were controlled by tuning the growth condition (growth temperature and time). In particular, the as-grown Bi₂O₂Se nanoribbons can be easily transfer onto arbitrary substrates (such as SiO₂/Si). The high electrical performance (high mobility, high current on/off ratio, and high on current density) of Bi₂O₂Se-nanoribbon-based transistors suggest that Bi₂O₂Se nanoribbons is a promising materials for fundamental investigations and high-performance electronic applications such as Gate-all-around FETs^{31,32} and tri-gate FETs³³.

Supporting Information: available free of charge via the internet at <http://www.whxb.pku.edu.cn>.

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